



# Chemical synthesis of optically active *cis*-cyclohexa-3,5-diene-1,2-diols and their 5-<sup>2</sup>H-derivatives

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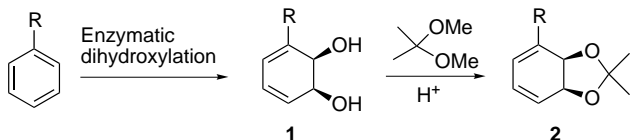
**Abstract**—A variety of optically active 3-substituted *cis*-cyclohexane-3,5-dien-1,2-diol acetonides were readily prepared from chiral 5-(*tert*-butyldimethylsilyloxy)-2-cyclohexenone through a seven-step reaction in good overall yield. The synthesis also allowed preparation of the acetonide having an <sup>2</sup>H-atom at the 5-position. © 2001 Elsevier Science Ltd. All rights reserved.

Enzymatic dihydroxylation of mono-substituted benzenes affords an efficient and practical method for preparing optically active cyclohexadienediols of the type **1** (Scheme 1). The compounds **1** and their hydroxy-protected derivatives, especially their acetonide **2**, have been widely utilized as starting material for synthesizing biologically active compounds, including natural products.<sup>1</sup> The methodology, however, with very few exceptions, allows production of only one enantiomer of **1**, and thus that of **2**, with the absolute stereochemistry shown in Scheme 1. We have now succeeded in developing an efficient and practical chemical synthesis of both enantiomers of **2**.<sup>2</sup>

We have recently introduced optically active 5-(*tert*-butyldimethylsilyloxy)-2-cyclohexenone (**3**) as a versatile chiral building block for synthesizing optically active cyclohexane derivatives.<sup>3</sup> The compound **3** was synthesized from commercially available, optically active ethyl 3-hydroxy-4-chlorobutyrate (**4**) via ethyl 3-(*tert*-butyldimethylsilyloxy)-5-hexenoate (**5**) by a six-

step reaction sequence in 51% overall yield as shown in Scheme 2 (via path a).

After the development of this synthesis, we investigated another approach to **5** starting from optically active epichlorohydrin (**6**), because **6** is commercially available at lower price in comparison with **4** and the enantiomeric excess (ee) of the commercially available **6** is somewhat higher than that of **4**.<sup>4</sup> As shown in path b in Scheme 2, the preparation of **5** from **6** can be readily carried out by the conventional reaction sequence. Thus, **6** was treated with vinylmagnesium bromide in the presence of a catalytic amount of CuCN in THF to afford 1-chloropent-4-en-2-ol, the chlorine of which was replaced by CN by reaction with potassium cyanide in methanol, and the resulting cyanohydrin was converted to **5** by acidic ethanolysis of the cyano group and the following silylation of the hydroxy group. The olefinic ester **5**, the ee of which was determined to be more than 98%, was conclusively prepared from **6** by a four-step reaction in 77% overall yield; thus, eventually, **3** was prepared from **6** by a seven-step reaction in 52% overall yield. As all the reagents used in either method for synthesizing **3** from **4** or **6** shown in Scheme 2 are readily available and inexpensive, and the reaction procedures are operationally simple, **3** can be easily prepared in quantity.

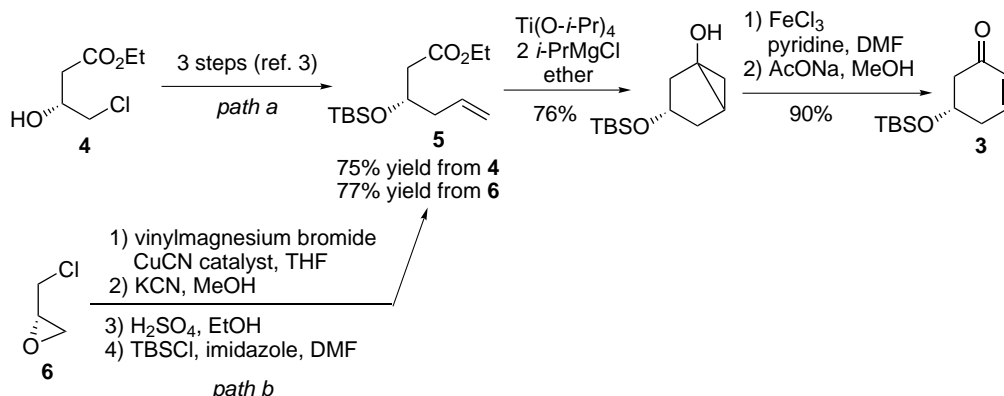


Scheme 1.

**Keywords:** *cis*-cyclohexane-3,5-dien-1,2-diol; optically active; chemical synthesis; deuterium.

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As a part of our continuing study to apply **3** as a chiral building block in asymmetric synthesis,<sup>3</sup> we have now found that a variety of optically active cyclohexadiene-diol acetonide **2** can be readily prepared from **3**. Although we used **3** with (*S*)-configuration in the present synthesis, as its antipode (*R*)-**3** is similarly



Scheme 2.

preparable because both enantiomers of **4** and **6** are, respectively, commercially available, the present method allows access to both enantiomers of **2**.

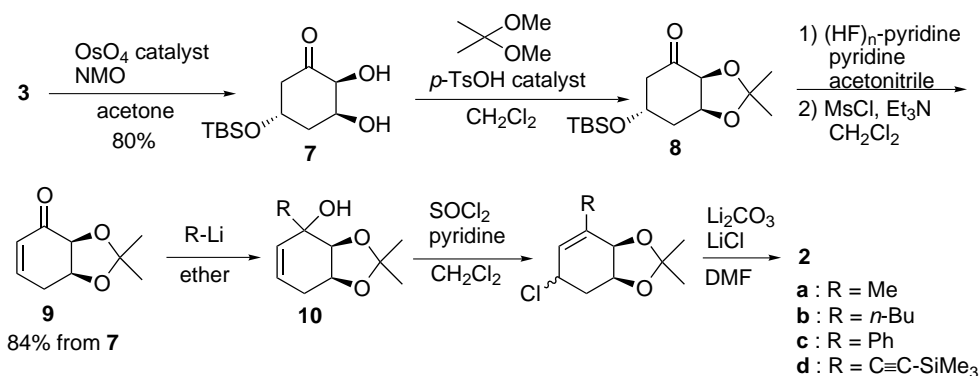
The preparation of **2** from **3** was carried out according to the procedure shown in Scheme 3. Thus, the dihydroxylation of **3** with catalytic osmium tetroxide–NMO gave **7** exclusively in 80% yield as previously reported,<sup>3c</sup> which in turn was converted quantitatively to the acetonide **8**, and the crude of which was used for the next reaction. Desilylation of **8** with (HF)<sub>n</sub>-pyridine and the following mesylation of the resulting hydroxy group with MsCl and Et<sub>3</sub>N was accompanied with β-elimination to furnish enone **9**<sup>5</sup> in 84% overall yield from **7**, thus, eventually, in 67% yield from **3**. From **9**, substituted cyclohexadienediol acetone derivatives **2a–d** could be prepared by a three-step reaction which involves 1,2-addition of the corresponding organolithium reagent to **9**, migrative chlorination of the resulting **10** with thionyl chloride, and the following elimination reaction using Li<sub>2</sub>CO<sub>3</sub>/LiCl<sup>6</sup> as a base. The yield of **2** from **9** and their [α]<sub>D</sub> values are summarized in Table 1. In conclusion, a variety of optically active acetone derivatives **2a–d** were prepared in 41, 47, 50, or 43% overall yield from **3**, respectively.<sup>7,8</sup> It is worth mentioning here that, at first, we were anxious about the successful conversion of **10** to **2** because further elimination of the resulting **2** to the corresponding phenol derivative was easily conceivable. Actually, the direct dehydration of **10** to **2** was unsuccessful. For example, treatment of **10** with *p*-TsOH

catalyst or with Tf<sub>2</sub>O/DBU gave a complicated mixture including the phenol derivative(s). Finally, to our delight, we were able to find the conditions enabling the conversion of **10** to **2** by the two-step procedure shown in Scheme 3; under these conditions, phenol derivative(s) was produced in less than 2% yield, if any.

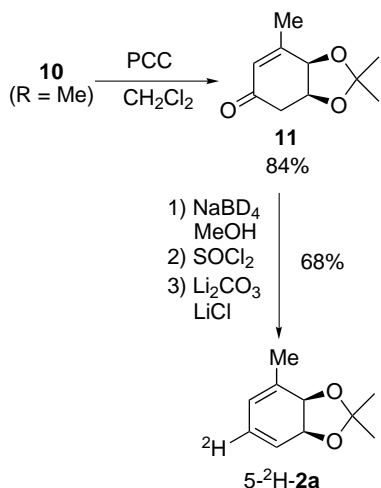
As many biologically important compounds have been prepared by using **2** as a starting material,<sup>1</sup> we next attempted to prepare **2** labeled with <sup>2</sup>H, which eventually allows access to those biologically important compounds labeled with <sup>2</sup>H. The preparation of **2** with <sup>2</sup>H at the 5-position was easily carried out as exemplified by the synthesis 5-<sup>2</sup>H-**2a** as shown in Scheme 4. Thus,

Table 1.

<b>2</b>	R	Yield from <b>9</b> (%)	[α] <sub>D</sub> Value
<b>2a</b> <sup>7,9</sup>	Me	62	[α] <sub>D</sub> <sup>20</sup> +93 ( <i>c</i> 0.27, MeOH) lit. <sup>7a</sup> [α] <sub>D</sub> <sup>20</sup> +93.7 ( <i>c</i> 2.98, MeOH)
<b>2b</b>	<i>n</i> -Bu	70	[α] <sub>D</sub> <sup>26</sup> +101 ( <i>c</i> 0.50, CHCl <sub>3</sub> )
<b>2c</b> <sup>7</sup>	Ph	75	[α] <sub>D</sub> <sup>26</sup> +204 ( <i>c</i> 0.12, CHCl <sub>3</sub> )
<b>2d</b>	Me <sub>3</sub> Si-C≡C	64	[α] <sub>D</sub> <sup>28</sup> +131 ( <i>c</i> 0.32, CHCl <sub>3</sub> )



Scheme 3.



Scheme 4.

PCC-oxidation of **10** (R=Me), derived from **9** and MeLi, afforded ketone **11** in 84% yield. The reduction of **11** with NaBD<sub>4</sub> and the successive treatment of the product with thionyl chloride and LiCO<sub>3</sub>/LiCl afforded the acetonide of 3-methyl-5-deutero-cyclohexa-3,5-dien-1,2-diol (5-<sup>2</sup>H-**2a**) with more than 93% of deuterium incorporation<sup>10</sup> in 68% overall yield.

### Acknowledgements

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- Compounds **4** with 97% ee and **6** with >98% ee are commercially available from Aldrich.
- $[\alpha]_D$  of **9** was  $[\alpha]_D^{26} +101$  (c 0.22, CHCl<sub>3</sub>) and the ee was found to be 97% by GC analysis with the use of a chiral column (Chirasil-DEX CB, 0.25 mm×25 m, Chrompack). Landais et al. reported the preparation of non-racemic **9** with  $[\alpha]_D^{25} +76.3$  (c 0.94, CHCl<sub>3</sub>) from chlorodimethylphenylsilane in 38% yield through a five-step reaction, which involves the Sharpless asymmetric dihydroxylation of 3-(hydroxydimethylsilyl)-1,4-cyclohexadiene: Angelaud, R.; Babot, O.; Charvat, T.; Landais, Y. *J. Org. Chem.* **1999**, 64, 9613–9624.
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- Compound **2b**: <sup>1</sup>H NMR  $\delta$  5.98 (dd,  $J=5.7, 9.6$  Hz, 1H), 5.78 (dd,  $J=3.9, 9.6$  Hz, 1H), 5.71 (d,  $J=5.7$  Hz, 1H), 4.65 (dd,  $J=3.9, 8.7$  Hz, 1H), 4.53 (d,  $J=8.7$  Hz, 1H), 2.14–2.31 (m, 2H), 1.25–1.58 (m, 4H), 1.39 and 1.41 (2s, each 3H), 0.92 (t,  $J=7.5$  Hz, 3H); <sup>13</sup>C NMR  $\delta$  138.7, 124.8, 122.3, 118.0, 105.2, 73.5, 71.3, 33.3, 29.4, 27.0, 25.2, 22.6, 14.1; IR (neat) 2958, 2878, 1603, 1458, 1369, 1258, 1048, 868, 804 cm<sup>-1</sup>. Anal. calcd for C<sub>13</sub>H<sub>20</sub>O<sub>2</sub>: C, 74.96; H, 9.68. Found: C, 75.22; H, 9.67%. Compound **2d**: <sup>1</sup>H NMR  $\delta$  6.35 (d,  $J=5.7$  Hz, 1H), 6.01 (dd,  $J=5.7, 9.6$  Hz, 1H), 5.92 (dd,  $J=3.3, 9.6$  Hz, 1H), 4.71 (dd,  $J=3.3, 8.1$  Hz, 1H), 4.54 (d,  $J=8.1$  Hz, 1H), 1.43 and 1.42 (2s, each 3H), 0.10 (s, 9H); <sup>13</sup>C NMR  $\delta$  130.6, 126.9, 123.3, 119.7, 105.9, 104.2, 98.8, 72.3, 71.4, 26.9, 25.3, 0.10; IR (neat) 2960, 2142, 1379, 1250, 1034, 844 cm<sup>-1</sup>. Anal. calcd for C<sub>14</sub>H<sub>20</sub>O<sub>2</sub>Si: C, 67.70; H, 8.12. Found: C, 67.38; H, 8.49%.
- Comparison of the  $[\alpha]_D$  value of **2a** thus prepared with the reported one indicates that no racemization occurred during conversion of **9** to **2**.
- Determined by <sup>1</sup>H NMR analysis.